ictly are needed for calculating the the Debye temperature, the Grünt, and the melting temperature as the are listed in Table I.

eisen γ and melting temperature of solid ous values of volume.

γ	T_m (°K)	
2.71	4.9	
2.49	6.6	
2.31	8.8	
2.17	11.7	
2.06	15.5	
1.96	20.8	174.114
1.87	28.0	
1.8	38.5	
1.74	53.9	
1.68	77.7	
1.65	94.5	

nt between the extrapolated experied by the methods outlined in this clusions:

we and isochores of the solid can be

point motion of the atoms composing ium is sufficient to account quite well ect of the zero point energy is taken calculations (*); and

ormula appears to be valid over an fact which deserves close theoretical

st principles, the U_0 -V relationship should. This has already been attempted with a really satisfactory theoretical interato-

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INTERVENTI E DISCUSSIONI

- С. Dомв:

Dr. Dugdale's observation that the θ values used in estimating the Lindemann constant C need to be revised would mean a far smaller variation of C with Λ^* than indicated in my paper. It may help to account for the apparent discrepancy noted between solid hydrogen and solid helium.